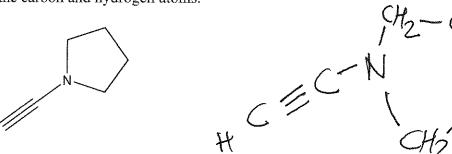
	Ley
Name	

pts

Quiz Ia February 22, 2017

1. Convert the following molecule from line bond notation to a Lewis structure that shows all the carbon and hydrogen atoms.





2. What is the official IUPAC name of the following molecule?

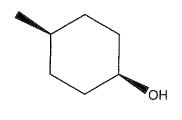
2 987854

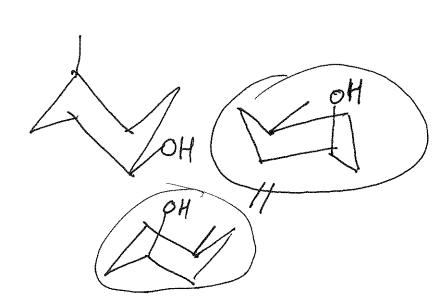
4,5-diethy/-7-methy/ decane

the question was more difficult than I intended

3. Draw both chair forms of the molecule shown below and circle the one that is lower in energy.

I will grade it more leniently





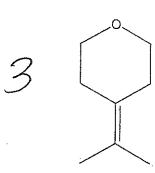
Name	Key	
Chiama		-

pts

Lab Instructor Belyayeva Chiemezie Gozde Khajo Mollica

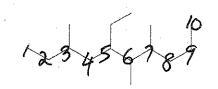
Quiz Ib February 22, 2017

1. Convert the following molecule from line bond notation to a Lewis structure that shows all the carbon and hydrogen atoms.



CH2 CH2 CH2 CH2

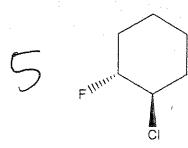
2. What is the official IUPAC name of the following molecule?

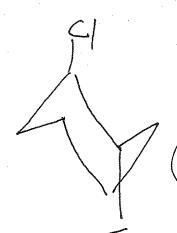


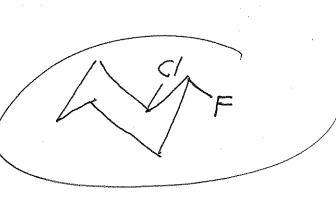
5 ethyl 3,6,7 trimethyl

decare

3. Draw both chair forms of the molecule shown below and circle the one that is lower in energy.





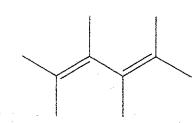


Name	

pts

Quiz Ic -February 22, 2017

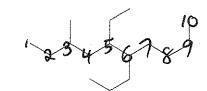
1. Convert the following molecule from line bond notation to a Lewis structure that shows all the carbon and hydrogen atoms.



CH3 CH3 CH3

2. What is the official IUPAC name of the following molecule?

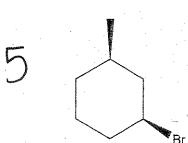


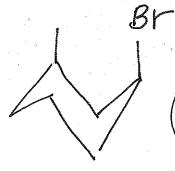


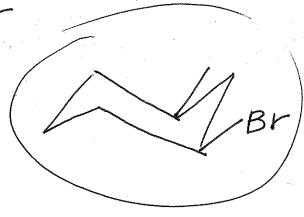
5 ethyl-6 propyl 3-methyl-56-dilly

decane

3. Draw both chair forms of the molecule shown below and circle the one that is lower in energy.







Name	

pts

Quiz IIa March 1, 2017

1. What is the formal charge on the C atom in the molecule shown below?

2. Two resonance forms of the same molecule are shown below. Draw arrows to show how the structure on the left would covert to the structure on the right and how the structure on the right would covert to the structure to the left.





- 3a. Draw a resonance structure for the molecule shown below.
- 3b. If your resonance structure has any formal charges, be sure to show them clearly.





pts

Quiz IIb March 1, 2017

1. What is the formal charge on the O atom in the molecule shown below?

2



2. Two resonance forms of the same molecule are shown below. Draw arrows to show how the structure on the left would covert to the structure on the right and how the structure on the right would covert to the structure to the left.

4

- 3a. Draw a resonance structure for the molecule shown below.
- 3b. If your resonance structure has any formal charges, be sure to show them clearly.

Name	

pt5

Quiz IIc March 1, 2017

1. What is the formal charge on the N atom in the molecule shown below?





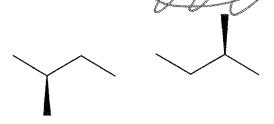
2. Two resonance forms of the same molecule are shown below. Draw arrows to show how the structure on the left would covert to the structure on the right and how the structure on the right would covert to the structure to the left.

- 3a. Draw a resonance structure for the molecule shown below.
- 3b. If your resonance structure has any formal charges, be sure to show them clearly.

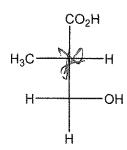
Name	

Quiz IIIa March 29, 2017

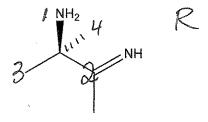
1. What is the relationship between the following two molecules? Are they identical) enantiomers, diastereomers or constitutional isomers?



2. Label all chiral centers in the molecule shown below.



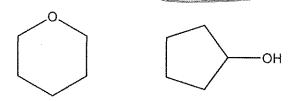
3. Determine the R/S configuration of the chiral center shown below. Make sure to clearly indicate the priority (1, 2, 3 or 4) of each group attached to the chiral carbon.



Name	

Quiz IIIb March 29, 2017

1. What is the relationship between the following two molecules? Are they identical, enantiomers, diastereomers or constitutional isomers?



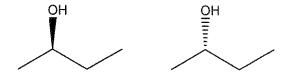
2. Label all chiral centers in the molecule shown below.

3. Determine the R/S configuration of the chiral center shown below. Make sure to clearly indicate the priority (1, 2, 3 or 4) of each group attached to the chiral carbon.

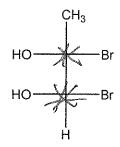
Name	

Quiz IIIc March 29, 2017

1. What is the relationship between the following two molecules? Are they identical enantiomers, diastereomers or constitutional isomers?



2. Label all chiral centers in the molecule shown below.

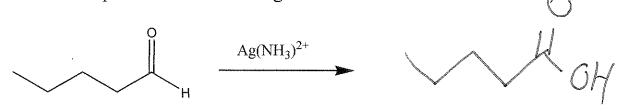


3. Determine the R/S configuration of the chiral center shown below. Make sure to clearly indicate the priority (1, 2, 3 or 4) of each group attached to the chiral carbon.

Name	

Quiz IVa April 5, 2017

1. Give the product of the following reaction:

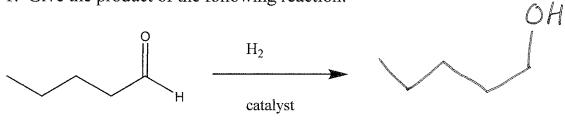


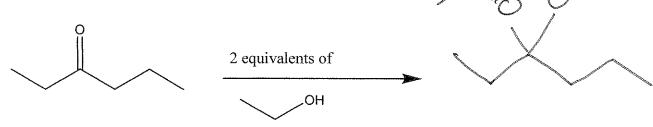
- 3a. Which functional group (acetal or hemiacetal) is shown in the molecule below?
- 3b. Draw the structures of the aldehyde and alcohol(s) that were used to synthesize this compound:

Name	

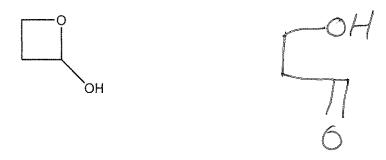
Quiz IVb April 5, 2017

1. Give the product of the following reaction:





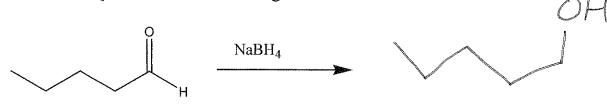
- 3a. Which functional group (acetal or hemiacetal) is shown in the molecule below?
- 3b. Draw the structures of the aldehyde and alcohol(s) that were used to synthesize this compound:



Name	

Quiz IVc April 5, 2017

1. Give the product of the following reaction:



- 3a. Which functional group (acetal or hemiacetal) is shown in the molecule below?
- 3b. Draw the structures of the aldehyde and alcohol(s) that were used to synthesize this compound:



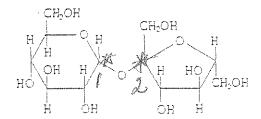
Name	

Quiz Va May 8, 2017

1. Convert the following sugar from the open form to the hemiacetal form. Draw a 5 membered ring.

Draw the anomeric carbon with the OH group α (alpha).

- 3. Analyze the following disaccharide:
 - a. Label the anomeric carbons with stars.
 - b. Which two carbons (give their numbers) link the two sugars together?
 - c. Is this compound a reducing sugar? In other words, does it undergo oxidation with Benedicts' or Tollens' reagent?



Name	

Quiz Vb May 8, 2017

1. Convert the following sugar from the open form to the hemiacetal form. Draw a 6 membered ring.

Draw the anomeric carbon with the OH group β (beta).

- 3. Analyze the following disaccharide:
 - a. Label the anomeric carbons with stars.
 - b. Is the linkage between the two sugars an alpha or beta linkage?
 - c. Is this compound a reducing sugar? In other words does it undergo oxidation with Benedicts' or Tollens' reagent?

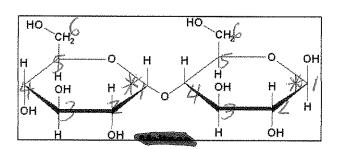
Name

Quiz Vc May 8, 2017

1. Convert the following sugar from the open form to the hemiacetal form. Draw a 6 membered ring.

Draw the anomeric carbon with the OH group β (beta).

- 3. Analyze the following disaccharide:
 - a. Label the anomeric carbons with stars.
 - b. Which two carbons (give their numbers) link the two sugars?
 - c. Is this compound a reducing sugar? In other words does it undergo oxidation with Benedicts' or Tollens' reagent?

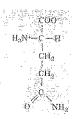




Name	

Quiz 6a May 15, 2017.

1. Does the following amino acid contain an acidic, basic polar or nonpolar side chain?

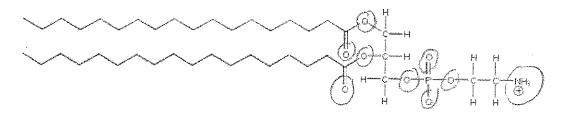


2. Draw the zwitterion form of the following amino acid:

3. Construct a dipeptide Asp Glu from the amino acids shown below. Make sure to draw the dipeptide in the form it would take in your body (at physiological pH):

American model oberes To

4. Below is a lipid found in a cell membrane. CAREFULLY circle the parts of the lipid that participate in hydrogen bonding.



Quiz 6h May 15, 2017

1. Does the following amino acid contain an acidic, basic, polar or nonpolar side chain?

2. Draw the zwitterion form of the following amino acid:

$$\begin{array}{c|c} \mathsf{O} & & & \\ \mathsf{II} & & & \\ \mathsf{H}_2 \mathsf{N} & & \mathsf{C} \mathsf{H} - \mathsf{C} & - \mathsf{O} \mathsf{H} \\ & \mathsf{I} & & \\ & \mathsf{H} & & \end{array}$$

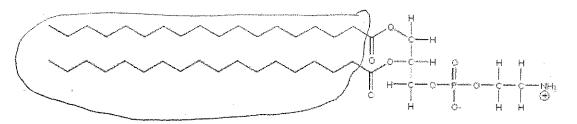
3. Construct a dipeptide Asn Asp from the amino acids shown below. Make sure to draw the dipeptide in the form it would take in your body (at physiological pH):

$$NH - C - CO_2$$

$$CH_2$$

$$CO_2$$

4. Below is a lipid found in a cell membrane. CAREFULLY circle the parts of the lipid that are nonpolar.



Name ____

Lab Instructor Belyayeva Chiemezie Yildirim Khajo Mollica

Quiz 6**£** May 15, 2017

1. Does the following amino acid contain an acidic, basic, polar or nonpolar side chain?

2. Draw the zwitterion form of the following amino acid:

3. Construct a dipeptide Gln Glu from the amino acids shown below. Make sure to draw the dipeptide in the form it would take in your body (at physiological pH):

4. Below is a lipid found in a cell membrane. CAREFULLY circle the parts of the lipid that are polar.

